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ORIGINAL ARTICLE

# Applying green analytical chemistry (GAC) for development of stability indicating HPLC method for determining clonazepam and its related substances in pharmaceutical formulations and calculating uncertainty



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## KEYWORDS

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Stability indicating method

**Abstract** Clonazepam contains one benzodiazepine ring in its chemical structure which makes it vulnerable to degradation. In this study, green analytical chemistry approach was applied in attempts for the development of validated stability indicating RP-HPLC method for determining clonazepam and its related substances in pharmaceutical formulation. Validation has been performed according to ICH guidelines. Assay was capable of simultaneous monitoring of the intact drug in the presence of its related substances within the same run. HPLC assay involved an ODS column and a mobile phase composed of 2% sodium dodecyl sulfate, 0.05 M sodium acetate buffer pH 3.5 and isopropanol in ratio (25:55:20) at a flow rate of 1.5 mL/min and detection was carried out at 254 nm. HPLC method allowed good resolution between the peaks that corresponded to the

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active pharmaceutical ingredients and its degradation products with good linearity, precision, accuracy, specificity, LOD and LOQ. The expanded uncertainty (0.33%) of the method was also estimated from method validation data. This analytical technique is not only ecofriendly but also faster than the conventional liquid chromatographic system official in the USP-36.

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## 1. Introduction

Due to scientific and public concern about the environment pollution, environmentally friendly practices have been introduced in different areas of society and research. In green analytical chemistry, sample preparation and LC analysis need special attention because hazardous solvents are often used. European medicine agency (EMA) mentioned that solvents like methanol and tetrahydrofuran are ranked by as hazardous solvents (ICH Topic Q3C (R4), 2009) and because of their inherent toxicity, safe detoxification of the waste solvents is essential, which may lead to high to very high disposal costs. Possibilities toward green LC include reducing solvent use, switching to more benign solvents and/or eliminating organic solvents (Armenta et al., 2008; Kerton and Marriott, 2013; Taylor, 2010; Clark and Tavener, 2006). Clonazepam [5-(2-Chlorophenyl)-7-nitro-3H-1,4-benzodiazepin-2(1H)-one] is mainly used as anticonvulsant, muscle relaxant and anxiolytic agent. Clonazepam is slightly soluble in acetone, chloroform, acetic anhydride, hardly soluble in methanol, isopropanol, ether, almost insoluble in water. In all of the aforementioned LC techniques, including the USP-36 relevant monograph, tetrahydrofuran and methanol have been used as a part of mobile phase and extraction procedure. The present investigation describes a rapid, accurate and precise RP-HPLC method for the determination of Clonazepam and its related compound A & B (Fig. 1) in pharmaceutical dosage forms within quality control laboratories without the need to use of hazardous, toxic solvents since this drug is being marketed in domestic and international market.

## 2. Experimental

### 2.1. Materials and chemicals

USP – reference standard materials of clonazepam (99.88%), related compound A and related compound B in addition to excipient used for preparing the placebo were kindly provided

by Sigma pharmaceutical industries (Egypt). All solvents and reagents were of HPLC grade, and were purchased from Sigma-Aldrich. HPLC grade water was obtained through a Milli-Q system (Millipore, Milford, USA) and was used to prepare all solutions.

Commercial samples of clonazepam tablets, manufactured by Sigma pharmaceutical industries – Egypt, were used in the applications.

### 2.2. Apparatus

Shimadzu HPLC system (Shimadzu, Kyoto, Japan), set to recycle the mobile phase and was equipped with a System Controller CBM-20A, Solvent Delivery Unit LC-20A, On-line Degassing Unit DGU-20A, and Photo-diode Array detector SPD-M20A. The peak areas were integrated automatically by computer using a Shimadzu LC solution V1.24 SP1 software program. A 20  $\mu$ L volume of sample was introduced into Auto-Sampler SIL-20A injector. Photo cabinet “Atlas Suntest CPS Plus” – Germany has been used for photodegradation.

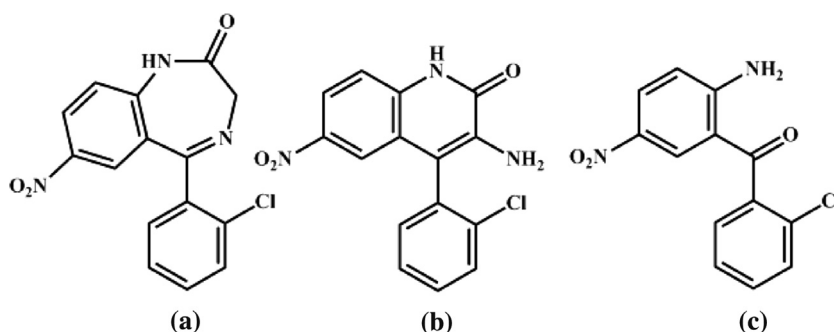
### 2.3. Analytical procedures

#### 2.3.1. Reference method

According to the relevant monograph in the USP-36 (USP, 2013) the elution was carried out on a BDS C<sub>8</sub> Hypersil column (150 mm  $\times$  4.6 mm, 5  $\mu$ m particle size) from Thermo scientific (Massachusetts, USA). All analyses were performed at ambient temperature under isocratic conditions with a mobile phase of ammonium phosphate buffer solution pH 8.0: methanol: tetrahydrofuran at a flow rate of 1.5 mL/min in ration (60:52:13, v/v), using UV detection at 254 nm. Diluent solution prepared by mixing water, methanol, and tetrahydrofuran in ration 60:52:13 was used for standard and sample preparation.

#### 2.3.2. Proposed liquid chromatography method

2.3.2.1. Preparation of stock solutions and standards. Stock standard solution of clonazepam, related compound A and



**Figure 1** Structures of clonazepam and its related compounds (a) clonazepam (b) clonazepam related A and (c) clonazepam related B.

related compound B was prepared in mobile phase at a concentration of 0.2 mg/ml. Series of dilutions at concentration levels of 4–140 µg/mL were obtained from stock solution by the appropriate dilution in mobile phase.

**2.3.2.2. Sample preparation.** Samples, of randomly selected clonazepam tablets, were dissolved and diluted to the appropriate volumes to get a concentration of 0.04 mg/ml using mobile phase as solvent. All samples were filtered through nylon sample filter (Whatman, 0.22 µm).

**2.3.2.3. Method development and chromatographic conditions.** A variety of mobile phases were investigated in the development of a stability-indicating HPLC method for the analysis of clonazepam and its related compound in pharmaceutical preparations. The suitability of mobile phase was decided on the basis of green analytical chemistry principles, selectivity, sensitivity of the assay, stability studies, and separation of clonazepam from the known related substances (A&B) beside the impurities formed during forced degradation studies. The elution was carried out on a BDS C<sub>8</sub> Hypersil column (250 mm × 4.6 mm, 5 µm particle size) from Thermo scientific (Massachusetts, USA). All analyses were performed at ambient temperature under isocratic conditions with a mobile phase of isopropanol: 2% sodium dodecyl sulfate (SDS): 0.05 M sodium acetate buffer (pH 3.5 ± 0.05) in ratio (20:25:55, v/v) at a flow rate of 1.5 mL/min, using DAD detector at 254 nm.

**2.3.2.4. Stability indicating capabilities of the proposed method.** The stability-indicating capability of the method was determined by its ability to separate the reference degradation substances (A&B) from the drug of interest in addition to subjecting reference solution of clonazepam (40 µg/mL) to forced degradation conditions by acidic, basic, oxidative, and photolytic conditions to evaluate the interferences in the quantitation of clonazepam. Standard solutions in 1 M hydrochloric acid and 1 M sodium hydroxide were used for the acidic and basic hydrolysis, respectively. Both solutions were kept at ambient temperature for 12 h. For oxidative degradation, solutions were prepared in hydrogen peroxide (3%) solution and kept at ambient temperature for 4 h. Photodegradation was induced by irradiating the neutral solution at 1.2 million lux hours for 24 h. The distance between the light source and the sample was maintained at 25 cm. These solutions were diluted with mobile phase to final concentration of 40 µg/ml and were injected into chromatographic system.

**2.3.2.5. Linearity.** Linearity was evaluated by determining the response of a series of dilutions of thirteen standard solutions from clonazepam and nine standard solutions of its related compounds (A&B) at a concentration range of 4–140 µg/mL and 4–64 µg/mL, respectively. Each dilution was injected in triplicate to plot the calibration curve. Slope, intercept, and regression coefficient ( $R^2$ ) of the calibration curves were calculated to ascertain linearity of the method.

**2.3.2.6. Precision.** For method repeatability, assay of authentic samples solutions was repeatedly performed six times on the same day (intra-day). For reproducibility, freshly prepared solutions at aforementioned concentration level were analyzed at different days (inter-day), and results were statistically evaluated in terms of % RSD. The authentic samples were

prepared by addition of the suitable amount of standard and the related compounds (A&B) to the placebo. These samples were handled as described under assay preparation to give the desired concentration.

**2.3.2.7. Accuracy.** Accuracy was calculated as the deviation of the mean from nominal concentration. To assess accuracy, freshly prepared placebo of the clonazepam pharmaceutical formulations was spiked with various amounts of clonazepam and its related compounds (A&B) to obtain the concentration levels of 30, 40 and 50 µg/ml. Each solution was injected in triplicate.

**2.3.2.8. Limit of detection (LOD) and limit of quantitation (LOQ).** The LOD is taken as the lowest concentration of an analyte in a sample that can be detected, but not necessarily quantified, under the stated conditions of the test while the LOQ is the lowest concentration of an analyte in a sample that can be determined with acceptable precision and accuracy under the stated conditions of test. They can be calculated from the linear calibration curve where LOD is the 3 times of residual standard deviation of the regression line (at the lower concentration) divided by the slope, while LOD is ten times the division product (Shrivastava and Gupta, 2011).

**2.3.2.9. Robustness.** In order to check the robustness, the effect of small but deliberate variations in the chromatographic conditions was evaluated. The conditions studied were flow rate (altered by ±0.01 mL/min), and pH of buffer solution (altered by ±0.1). These chromatographic variations were evaluated for retention time, peak area, number of theoretical plate and asymmetry factor for each clonazepam and its related compound.

**2.3.2.10. Selectivity and system suitability.** Selectivity of this method was indicated by the absence of any excipient interference at the retention times of the peaks of clonazepam and its related compounds with an acceptable resolution in the system suitability solution. The absence of interfering peak was evaluated by injecting a blank sample consisting of diluent and placebo. The test is only valid if the resolution between clonazepam related A and clonazepam related B is more than 2 as stated in the relevant monograph in the USP 36 (USP, 2013).

**2.3.2.11. Uncertainty of the method.** Among the different sources of uncertainty, the uncertainty associated with calibration appears to be the most important source in the overall uncertainty (ICH guideline Q2B, 2005).

The quantification of clonazepam by was assess through the calibration curve equation ( $Y = aX + b$ ), where  $Y$  is the response,  $a$  is the slope,  $X$  is the concentration of clonazepam and  $b$  is the intercept. Based on this information, the uncertainty of HPLC's result was estimated by the following equation:

$$U_{A\%} = A\% \times t_{1-\alpha/n-2} \times \sqrt{\frac{(\sigma_y + \sigma_b)^2}{(Y - b)^2} + \left(\frac{\sigma_a}{a}\right)^2}$$

where  $U_{A\%}$  is the expanded uncertainty,  $t_{1-\alpha/n-2}$  is the  $t$ -student for confidence level of  $1-\alpha$  and  $n-2$  degrees of freedom ( $n$  correspond to the numbers of standard employed to obtain

calibration curve),  $A\%$  is the result of sample (in percentage),  $\sigma Y$  is the standard deviation of the areas obtained in chromatograms of sample,  $Y$  is the average of the areas obtained in chromatograms of sample,  $\sigma b$  is the error for the intercept and  $\sigma a$  is the error for the slope.

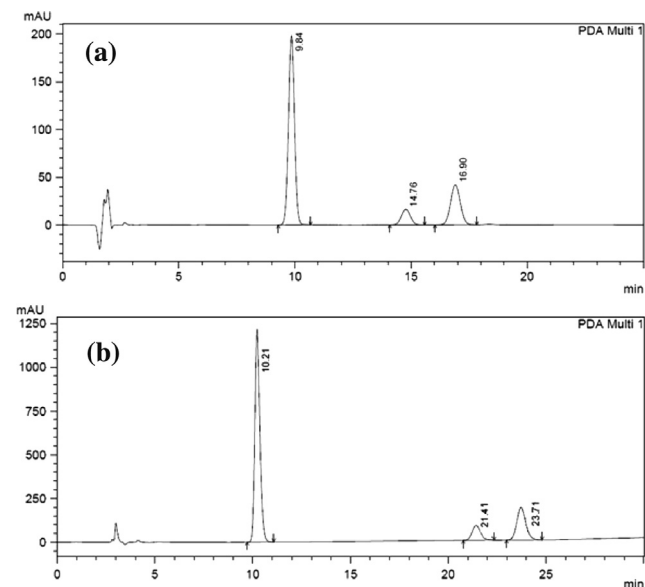
### 3. Results and discussions

#### 3.1. Optimization of the chromatographic conditions

The HPLC procedure was optimized with a view to develop an ecofriendly stability-indicating method. No internal standard was used because no extraction or separation step was

involved. Considering the green chemistry fundamentals (Clark et al., 2006), various solvent systems as mobile phases were tried for the development of a green and environmentally benign HPLC method. Of several solvents and solvent mixtures investigated, the mobile phase of isopropanol: 2% SDS: 0.05 M sodium acetate buffer (pH  $3.5 \pm 0.05$ ) in ratio (20:25:55, v/v) was found to furnish sharp, well defined peaks with very good symmetry (around 1) and appropriate separation (NLT 2) between the peak of interest and the related compounds as a system suitability requirement in the relevant monograph in the USP-36 (Fig. 2).

Other organic modifiers and surfactants were tried such as Brij-L23, tween 20 and tween 80 did not resulted in chromatographic system as good as proposed method.



**Figure 2** Separation of system suitability solution using (a) the proposed green method and (b) the USP-36 method (Clonazepam at 9.84, related compound A at 14.86 and related compound B at 16.90) and (b) the USP-36 method (Clonazepam at 10.21, related compound A at 21.41 and related compound B at 23.71).

#### 3.2. Method validation

##### 3.2.1. Linearity

The calibration curves of clonazepam, clonazepam related A and clonazepam related B were evaluated by the linear least square analysis. The calibration curve plotted between concentration of drug and peak area and the regression equations were calculated and the correlation coefficients of the calibration curves were found to be  $R^2 > 0.999$  in all samples types. Linearity was established in concentration range of 4–140  $\mu\text{g/ml}$  for clonazepam and 4–64  $\mu\text{g/ml}$  for clonazepam related compounds A & B as reported in Table 1. The linear regression data for the calibration plot were indicative of a good linear relationship between peak area and concentration over a wide range. The correlation coefficient was indicative of high significance. The intercept of the ordinate showed the calibration plot did not deviate from linearity.

##### 3.2.2. Accuracy

Accuracy was determined by the standard addition method. Three sets of placebo preparations were spiked with 32, 40 and 48  $\mu\text{g/ml}$  of clonazepam and related substances reference standards and the mixtures were analyzed by the proposed method. The experiment was performed in triplicate. Good

**Table 1** Linearity of calibration curve for clonazepam and its related substances.

Clonazepam		Clonazepam related compound A		Clonazepam related compound B	
Concentration ( $\mu\text{g/ml}$ )	Peak area	Concentration ( $\mu\text{g/ml}$ )	Peak area	Concentration ( $\mu\text{g/ml}$ )	Peak area
4	354,436	4	43,266	4	116,673
8	714,285	8	87,160	8	240,323
16	1,438,491	16	176,453	16	493,393
24	2,153,360	24	273,040	24	739,427
32	2,880,450	32	367,352	32	978,949
40	3,594,755	40	451,953	40	1,232,962
48	4,323,766	48	545,139	48	1,475,217
56	5,042,151	56	638,419	56	1,715,512
64	5,775,803	64	734,168	64	1,926,860
80	7,304,113	$R^2$	0.9998	$R^2$	0.9997
100	9,080,737	Slope	11496.851	Slope	30429.064
120	10,822,764	Intercept	4458.937	Intercept	3781.023
140	12,556,717				
$R^2$	0.9999				
Slope	90217				
Intercept	231.86				

**Table 2** Estimation of the accuracy as an item for validation of the proposed HPLC method for the determination of clonazepam and its related substances.

	Amount taken ( $\mu\text{g}$ )	Amount found ( $\mu\text{g}$ )	Percent recovery	Mean percentage recovery	RSD%
Clonazepam	32	31.87	99.59%	99.57%	0.11
	40	39.78	99.45%		
	48	47.84	99.67%		
Clonazepam related compound A	32	32.28	100.88%	99.77%	0.97
	40	39.62	99.05%		
	48	47.71	99.40%		
Clonazepam related compound B	32	32.16	100.50%	100.97%	0.42
	40	40.53	101.33%		
	48	48.52	101.08%		

**Table 3** Repeatability and intermediate precision for clonazepam and its related substances.

Exp. No.	Repeatability on day 1			Repeatability on day two		
	Clonazepam	Related A	Related B	Clonazepam	Related A	Related B
1	99.90%	100.06%	100.92%	100.00%	100.16%	101.02%
2	99.93%	100.33%	101.16%	100.23%	100.63%	101.46%
3	99.96%	100.30%	100.99%	100.16%	100.50%	101.19%
4	99.97%	100.30%	101.76%	100.37%	100.70%	102.17%
5	99.79%	100.27%	101.91%	99.89%	100.37%	102.01%
6	99.93%	100.94%	101.65%	100.13%	101.14%	101.86%
Mean	99.91	100.37	101.40	100.13	100.58	101.62
SD	0.064	0.298	0.427	0.168	0.334	0.464
RSD (%)	0.064	0.297	0.421	0.168	0.332	0.456

recoveries (101.33–99.40%) were obtained at each concentration level with a very low % RSD (0.97–0.11%) which indicated the accuracy of the proposed method (Table 2).

### 3.2.3. Precision

**3.2.3.1. Analysis repeatability.** It was evaluated by carrying out the analysis of the six homogenous solutions of same test sample and content of related substances. The determinations were carried out one after the other under conditions as similar as possible. The relative standard deviation was calculated from the results of the obtained observations (Table 3).

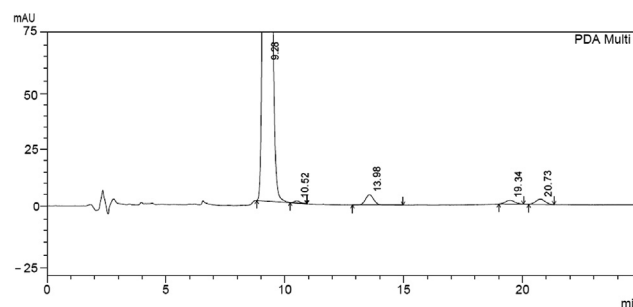
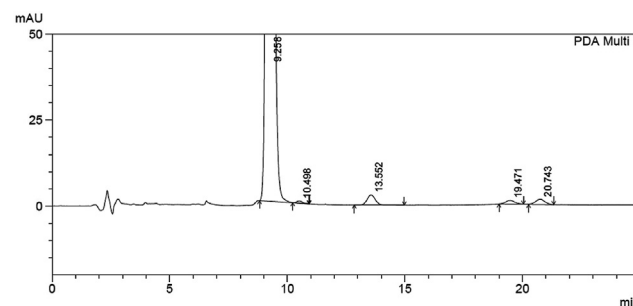
**3.2.3.2. Intermediate precision.** The intermediate precision of the method was checked by determining precision on a different instrument, analysis being performed in different laboratory on a different day. The relative standard deviation was calculated from the results of the obtained observations.

In all cases the RSD was lower than 2 (Table 3).

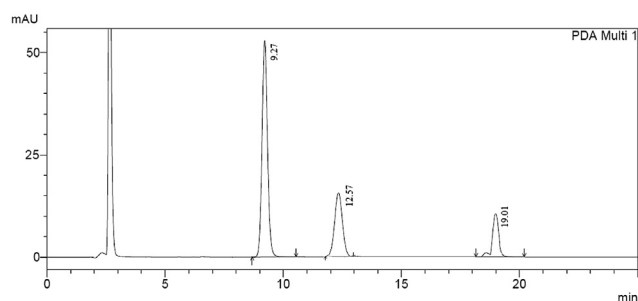
### 3.2.4. Specificity

The specificity of the method was evaluated through its ability to discriminate between the peak of the parent drug and those due to its related compounds A & B (Fig. 2).

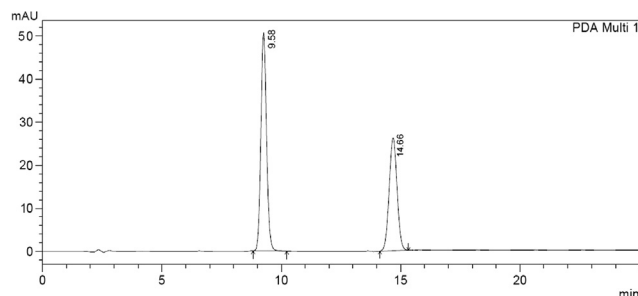
In the same regard a forced degradation study was established, clonazepam stock solution was exposed to various stress conditions. On treating with 5 N NaOH extensive degradation has been occurred, when treated with 1 N NaOH for one hour, the height of peak was reduced and four new peaks of degradation products were observed (Fig. 3). Almost the same has been happened on treating with 1 N HCl for one hour (Fig. 4).

**Figure 3** Clonazepam solutions under basic degradation conditions (Clonazepam at 9.28, basic degradation products at 10.52, 13.98, 19.34 and 20.73).**Figure 4** Clonazepam solution under acidic degradation conditions (Clonazepam at 9.26, acidic degradation products at 10.50, 13.55, 19.47 and 20.74).





**Figure 5** Clonazepam solution under oxidative degradation conditions (Clonazepam at 9.27, oxidative degradation products at 12.75 and 19.01).



**Figure 6** Clonazepam solution under photo-irradiation conditions (Clonazepam at 9.58, photo degradation product at 14.66).

Clonazepam was found to degrade more rapidly in oxidative conditions. Upon reflux with 30% hydrogen peroxide for 1 h, peak height was reduced, with the appearance of two new degradation peaks (Fig. 5). One degradation peak has been separated with dramatic change in the peak of the parent drug has been happened on exposing to photo-irradiation conditions in photo cabinet (Fig. 6). The above results show the method is stability indicating.

### 3.2.5. Detection (LOD), quantification (LOQ) limits and system suitability

Limit of detection (LOD) and the limit of quantitation (LOQ) were calculated as 3 S.D. and 10 S.D. of the blank, respectively, according to the treatment by Miller and Miller (Sadeghi et al., 2013). Before injecting solutions, the column was equilibrated for at least 20 min with the mobile phase. Their values prove the sensitivity of the proposed method and that it can be used for detection and quantification of trace amounts of clonazepam and its related compounds. The obtained system suitability parameters were satisfactory according to the relevant USP-36 monograph (Table 4).

### 3.2.6. Robustness

The robustness of an analytical procedure, as defined by the ICH refers to its capability to remain unaffected by small and deliberate variations in method parameters (Potka et al., 2013). Robustness was determined by recording the effect of slight change on the separation conditions of buffer and pH. Table 5 shows that the theoretical plates, tailing factor and resolution of clonazepam and its related substances are not changed significantly which indicates the proper robustness of the method.

### 3.2.7. Uncertainty estimation

The expanded uncertainty amount (0.33% at confidence level of 95%), was estimated based on the error of the slope and the intercept. Considering HPLC method, a six standard calibration curve may decrease the uncertainty estimated, as it decrease the *t*-student multiplier. Increasing the number of injections per standard level may also contribute in the reduction of final uncertainty (De Melo Abreu et al., 2005; Lourenço, 2012). Other sources of uncertainties, e.g. uncertainties of volumetric glassware, were not considered due to its non significant contribution into the total budget of the uncertainty.

**Table 4** LOD, LOQ and system suitability parameters\*.

Parameters	Clonazepam	Clonazepam related compound A	Clonazepam related compound B
LOD (µg/ml)	0.024	0.386	0.357
LOQ (µg/ml)	0.0799	1.2875	1.1893
Theoretical plates (h)	18625.64	8854.82	9570.59
Tailing factor (T)	1.104	0.988	1.032
Resolution	—	6.865	2.815

\* LOD, limit of detection; LOQ, limit of quantitation.

**Table 5** Evaluation of robustness of the proposed method.

Drug name	Variations	Chromatographic parameters			
		Retention time	Area	Theoretical plates	Asymmetry
Clonazepam	Change in buffer pH	9.83	3594745	18620.52	1.103
	Flow rate at 1.51 ml/min	9.75	3594760	18624.83	1.102
Clonazepam related compound A	Change in buffer pH	14.66	451943	8840.77	0.998
	Flow rate at 1.51 ml/min	14.56	451961	8857.94	1.009
Clonazepam related compound B	Change in buffer pH	16.85	1232974	9511.48	1.026
	Flow rate at 1.51 ml/min	16.75	1232950	9514.59	1.031

### 3.3. Application

Samples of clonazepam 0.5 mg tablets ( $n = 6$ ) were analyzed for the parent drug and its related substances by this method and the results showed a percent recovery of 100.1% and a RSD of 0.60% for 6 replicates while the related substances were within the accepted limits. The method was used for analyzing samples of the stability program of the drug product also for analysis of raw materials samples which give results between 99.6% and 100.2%.

### 4. Conclusion

The proposed RP-HPLC method is simple, selective, rapid, accurate, precise, reproducible, robust, sensitive and stability indicating. Therefore, the method was applied to the assay of clonazepam in tablets dosage form. The method is also simple in terms of sensitivity, use of an environmentally benign mobile phase, simple extraction procedures, relatively lower retention time and no internal standard is required. These advantages make this method superior for the routine analysis of clonazepam and its related substances in commercially available formulations. The method could also be applied for the prediction of shelf life in the stability studies of the mentioned formulations because it is having stability-indicating properties. The replacement of widely-used hazard solvents and chemicals with new, innocuous, and less toxic ones provides environmentally benign alternatives to the more hazardous chemicals and processes in the field of drug/pharmaceutical analysis. Finally, estimation of the uncertainty is important fulfill one of the ISO 17025 requirements aiming to guarantees the quality and reliability of the obtained results.

### Authors' contribution

Abdallah Shalaby and Ahmed Badr Eldin participated in the work conception, design, and coordination, helped to draft the paper and revised it critically for important intellectual content. Mahmoud Abdallah, Moataz Shaldam and Mohamed

Abdallah participated in the design and carried out the HPLC analysis and wrote the paper. All authors read and approved the final paper.

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